



CODEN [USA]: IAJPBB

ISSN: 2349-7750

**INDO AMERICAN JOURNAL OF  
PHARMACEUTICAL SCIENCES**<http://doi.org/10.5281/zenodo.1135275>Available online at: <http://www.iajps.com>

Research Article

**ACHYRANTHES ASPERA MEDIATED GREEN SYNTHESIS OF  
SILVER NANOPARTICLES****K. Ashok, K. Sivakumari\* and S. Rajesh**

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**Abstract**

*Bio-inspired AgNPs were rapidly synthesized at room temperature using fresh aqueous leaf extract of Achyranthes aspera. A green and low cost synthesis was effective in the formation of stable crystalline NPs in the solution. Carboxylic, ketone and aldehyde groups present in the A. aspera leaf extract functioned as reducing as well as stabilizing agent to produce shape controlled AgNPs. SPR confirmed the formation of AgNPs in UV-Visible spectra at 450 nm. The XRD result also showed the presence of elemental Ag<sup>+</sup> as a crystalline nature and the FT-IR analysis was carried out to identify and study the functional groups responsible for the bioreduction of Ag<sup>+</sup>. TEM and SEM with EDX image showed spherical crystalline AgNPs. Thermogravimetric analysis was used to measure the weight loss of AgNPs as a function of temperature under a controlled atmosphere. Hence, the plant-based bio AgNPs could be used in biomedical applications.*

**Keywords:** *Achyranthes aspera, Biosynthesis, AgNPs***Corresponding author:**

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Please cite this article in press as *K. Sivakumari et al., Achyranthes Aspera Mediated Green Synthesis of Silver Nanoparticles., Indo Am. J. P. Sci, 2018; 05(01).*

**INTRODUCTION:**

Nanomedicine is an emerging field expanding rapidly because of the development and incorporation of new nano composites into a range of products and technologies. In recent years, the application of nanoparticles (NPs) in medicine has increased and expanded to the fields of molecular imaging [1], drug delivery [2], diagnosis and treatment of cardiovascular diseases [3], wound healing [4] and development of materials and medical devices with antimicrobial properties [5].

New applications of NPs and nanomaterials are emerging rapidly in biomedical sciences [6]. This decade has witnessed the inception of new significant technological products particularly based on nanotechnology; NPs synthesis is being widely explored, since they exhibit unique size and shape dependent properties for applications in optics, electronics, catalytic systems, magnetic and biomedical fields such as HIV inhibition, cancer cell cytotoxicity and genotoxicity [7]. Apart from this, recently the anti-tumor effect of AgNPs has been reported against different cancerous cell lines [8]. NPs with the size range between 1 and 1000 nm are mainly explored for the diagnosis and treatment of human cancers, which led to the new discipline of nano-oncology [9].

There are number of methods used for the synthesis of silver nanoparticles (AgNPs) including physical and chemical methods [10-13], electrochemical reduction [14-15], photochemical reduction [16] and thermal evaporation [17-18]. However, rapid and green synthesis method using plant extract has developed enormous interest in AgNPs synthesis due to green chemistry approach. Moreover, it is simple, cost effective, eco-friendly, easily scaled up for large scale synthesis, without using toxic and redundant chemicals in solid, liquid and gaseous form [19]. Indeed, a number of bacteria [20], fungi and yeast have been well-known for synthesis of non-toxic noble NPs [21]. However the microbial-mediated synthesis of NPs is not industrially feasible as it requires expensive medium and maintenance of highly aseptic conditions [22].

In this context, plant-mediated NPs synthesis seems to be a cost-effective as well as eco-friendly method. Moreover, NP synthesis from plants with medicinal properties proves to be beneficial in treating various ailments in a better and easy way. On such plant is *Achyranthes aspera* L. (Amaranthaceae), which is distributed as weed throughout India, tropical Asia and other parts of the world. Ayurvedic, Yunani practitioners and Kabirajes use different parts of this

plant to treat leprosy, asthma, fistula, piles, arthritis, wound, insect and snake bite, renal and cardiac dropsy, kidney stone, diabetes, dermatological disorders, gynecological disorders, gonorrhoea, malaria, pneumonia, fever, cough, pyorrhoea, dysentery, rabies, hysteria, toothache *etc.* The plant is a popular folk remedy in traditional system of medicine throughout the tropical Asian and African countries. The plant is reported to be used as antimicrobial, larvicidal, antifertility, immunostimulant, hypoglycemic, hypolipidemic, anti-inflammatory, antioxidant, diuretic, cardiac stimulant, antihypertensive, antianasacra, analgesic, antipyretic, antinoiceptive, prothyroidic, antispasmodic and hepatoprotective.

Phytochemical investigations were carried out on this plant by several authors, which revealed the presence of sterols, alkaloids, saponins, sapogenins, cardiac glycosides, ecdysterone *etc.*, from different parts of this plant. Some other species of the genus *Achyranthes* viz. *A. fauriei*, *A. bidentata*, *A. japonica*, *A. ferruginea* *etc.* have also been investigated for their active constituents and pharmacological potential [23-33]. Survey of literature revealed that NPs synthesis from this plant is scanty. In view of this, the present study was designed to biosynthesize NPs from *Achyranthes aspera* leaves to study the reducing Ag<sup>+</sup> ions and stabilizing the particles and confirm AgNP synthesis by using various spectroscopy and microscopic methods.

**MATERIALS AND METHODS:*****Collection and identification***

Fresh leaves of *Achyranthes aspera* were collected from Presidency College, Chennai, Tamil Nadu, India, and were authentically identified by Central Council for Research in Ayurveda and Siddha, Chennai, India, as *Achyranthes aspera*. (Amaranthaceae) with voucher specimen no: 16475.

***Preparation of leaf extract and synthesis of AgNPs***

Twenty grams of fresh leaves were washed thoroughly in tap water and distilled water for 30 min. in order to remove the debris. The aqueous extract was prepared by taking 25 g of washed and finely chopped leaves in 100 ml conical flask along with 100 ml of distilled water and the mixture was boiled at 45°C for 30 min. This aqueous extract was filtered through Whatmann No. 1 filter paper and was used for synthesis of AgNPs. 10 ml of this aqueous leaf extract was added to 90 ml of 1mM aqueous silver nitrate solution for the synthesis of AgNPs. A control setup silver nitrate was also maintained without *A. aspera* extract.

**Qualitative phytochemical analysis**

Preliminary phytochemical analysis of aqueous extract was carried out by method of Harborne (1973) and Parekh and Chanda (2007) [34-35].

**Characterization of silver NPs**

UV-Visible spectra were recorded as a function of the reaction time on PG Instruments spectroscopy. The studies on size, morphology and composition of the NPs were performed by means of TEM (PHILIPS TECNAI 10) and SEM with EDX (Carl Zesis). The purified AgNPs were examined for the presence of biomolecules using FTIR analysis. Briefly, the spectrum obtained from the dried sample was recorded on FT-IR spectrum (Perkin-Elmer, USA) in the diffuse reflectance mode at a resolution of 4 cm<sup>-1</sup> in KBr pellets. Particle size analyzer was done dynamic light scattering (Malvern, MAL 1062727, UK). Crystalline AgNPs were determined by XRD. Briefly, the biosynthesized AgNPs were laid onto glass substrates on Phillips PW 1830 instrument operating at a voltage of 40 kV and current of 30 mA

with Cu K $\alpha$ 1 radiation. Thermal stability of NPs was done by TG-DTA (NETZSCH STA 409 PC/PG).

**RESULTS AND DISCUSSION:**

The present study was aimed to identify the phytochemicals present in *Achyranthes aspera* and to synthesize AgNPs from the aqueous extract of *A. aspera* leaves which is distributed as weed throughout India. Phytochemical study of *A. aspera* leaf extract shows the positive results for carbohydrates, tannins, phenols, flavonoids and triterpenoids. On the other hand, acids, alkaloids, anthocyanins and betacyanins, cardiac glycosides, coumarins, glycosides, proteins, quinones, saponins, starch and steroids were absent in the aqueous extract (Table 1). The bioactive compounds such as polyphenol, carbohydrates, vitamin and trace elements present in the leaf extract plays an important role as an antioxidant, anticancer, antitumor, anti-inflammatory, anti-obesity, anti-helminthic, analgesic, anti-pyretic, anti-ociceptive, anti-hepatitis, hepatoprotective, cardiac and Diuretic agent [36-40].

**Table 1: Qualitative phytochemical analysis of *A. Aspera* aqueous extract**

| S.No. | Phytochemicals               | Aqueous extract |
|-------|------------------------------|-----------------|
| 1.    | Acids                        | -               |
| 2.    | Alkaloids                    | -               |
| 3.    | Anthocyanins and Betacyanins | -               |
| 4.    | Carbohydrates                | +++             |
| 5.    | Cardiac Glycosides           | -               |
| 6.    | Coumarins                    | -               |
| 7.    | Flavonoids                   | +++             |
| 8.    | Glycosides                   | -               |
| 9.    | Phenols                      | +++             |
| 10.   | Proteins                     | -               |
| 11.   | Quinones                     | -               |
| 12.   | Saponins                     | -               |
| 13.   | Starch                       | -               |
| 14.   | Steroids                     | -               |
| 15.   | Tannins                      | +++             |
| 16.   | Terpenoids                   | ++              |
| 17.   | Triterpenoids                | +++             |

+++ Strongly present,  
+ Present

++ Mildly present  
- Absent

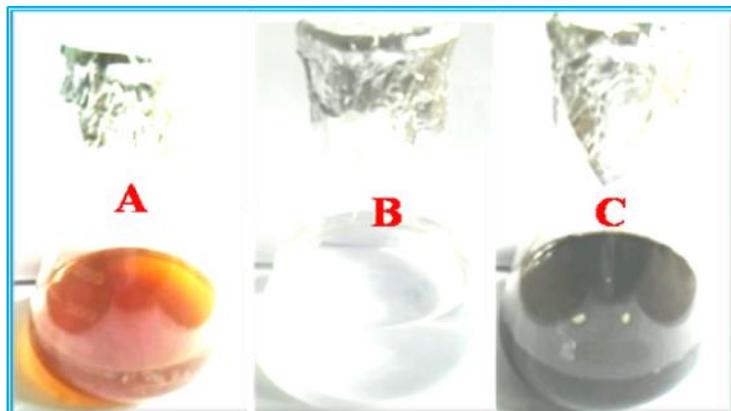


Fig. 1: A. Aqueous extract of *A. aspera* (pale yellowish brown colour)

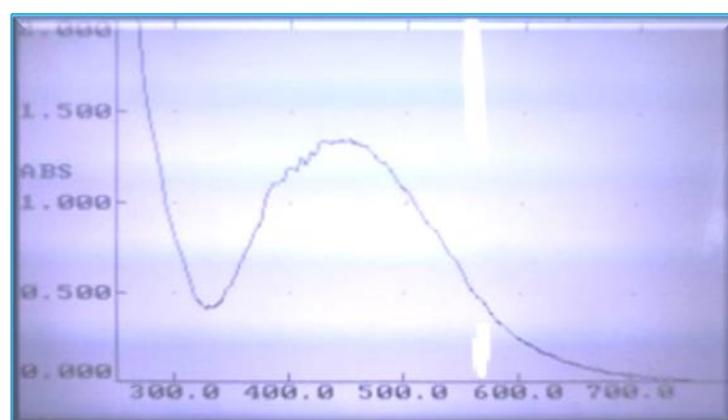


Fig. 2: UV-Vis spectral image of aqueous extract-based synthesized AgNPs.

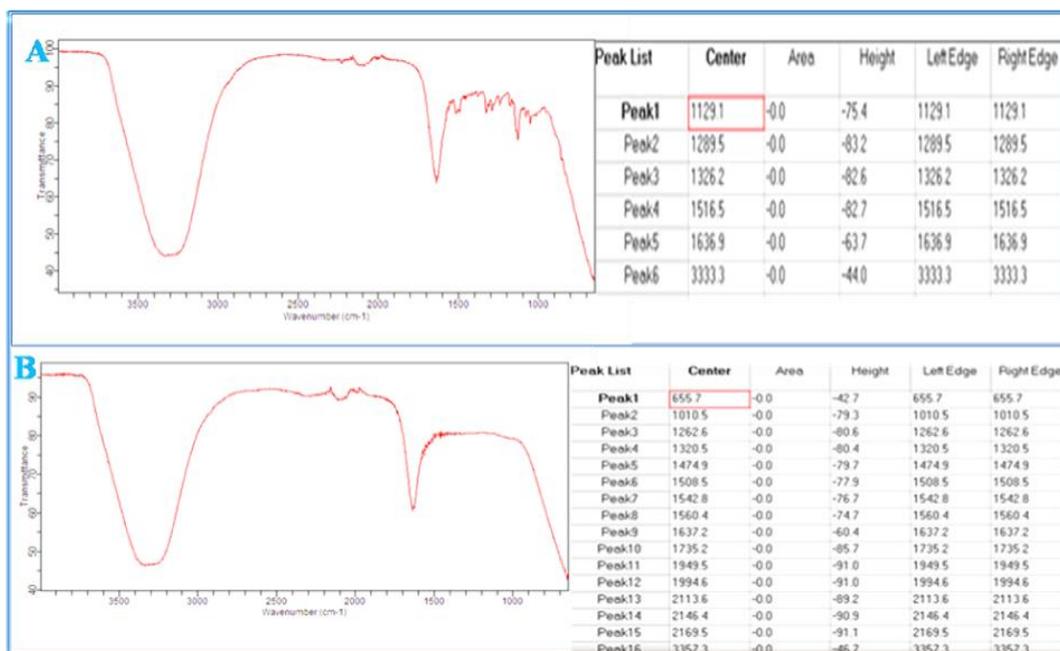


Fig. 3: FT-IR spectral image of various functional groups (1000 to 3500 cm<sup>-1</sup>) A- *A. aspera* aqueous extract and B- Aqueous extract-based synthesized AgNPs

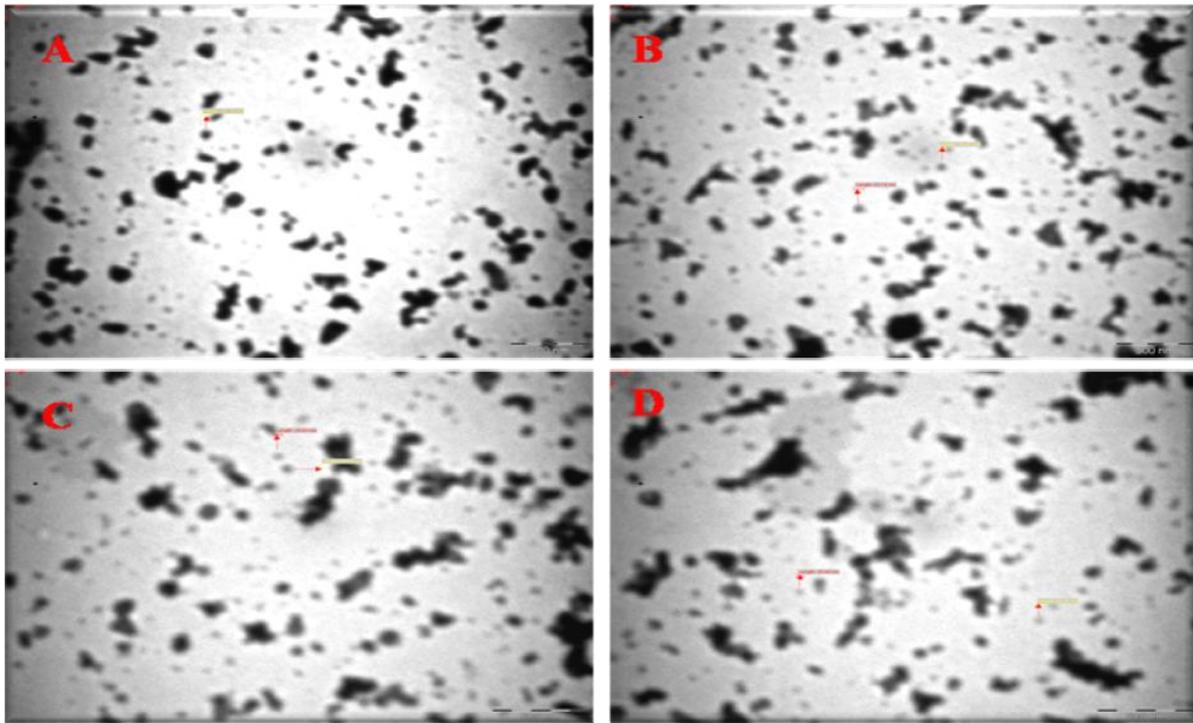


Fig. 4: TEM images of AgNPs formed by reduction of silver nitrate using *A. aspera* (A-D) 500 nm

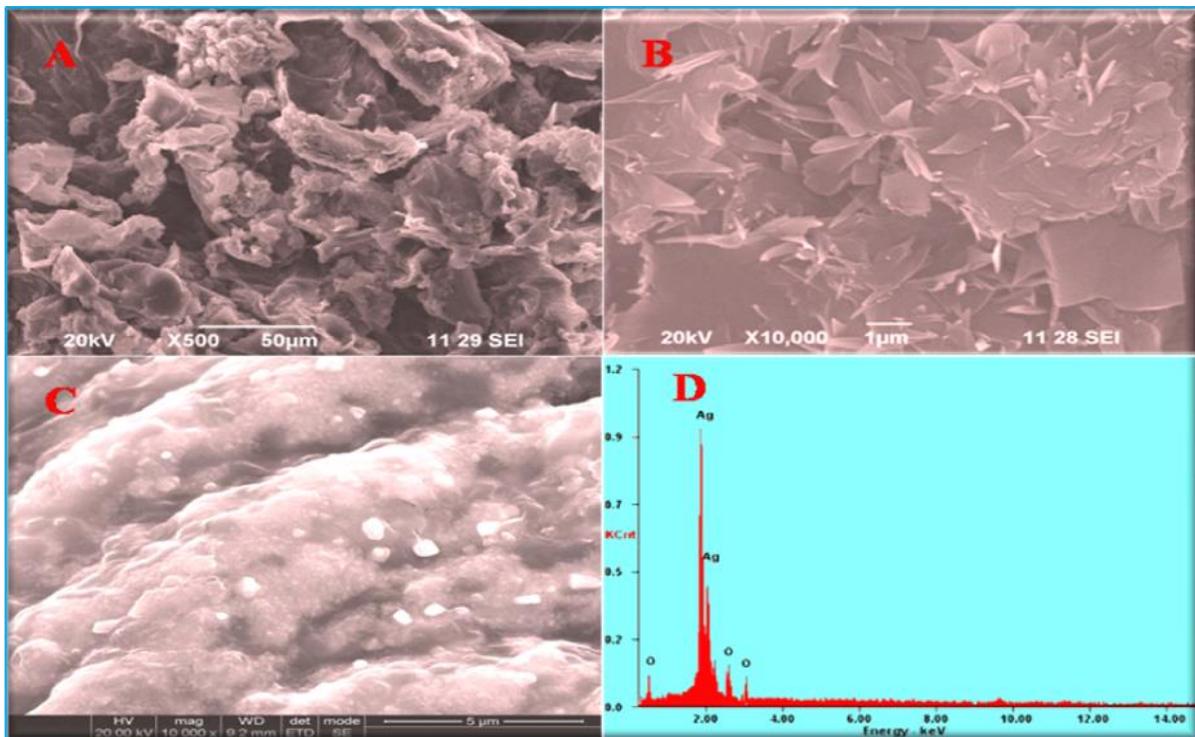


Fig. 5: A. SEM micrograph of *A. aspera* aqueous extract, B. SEM micrograph of 1 mM silver nitrate solution, C. SEM micrograph of *A. aspera* aqueous extract-based synthesized AgNPs and D. EDX analysis of AgNPs synthesized from the aqueous extract of *A. aspera*

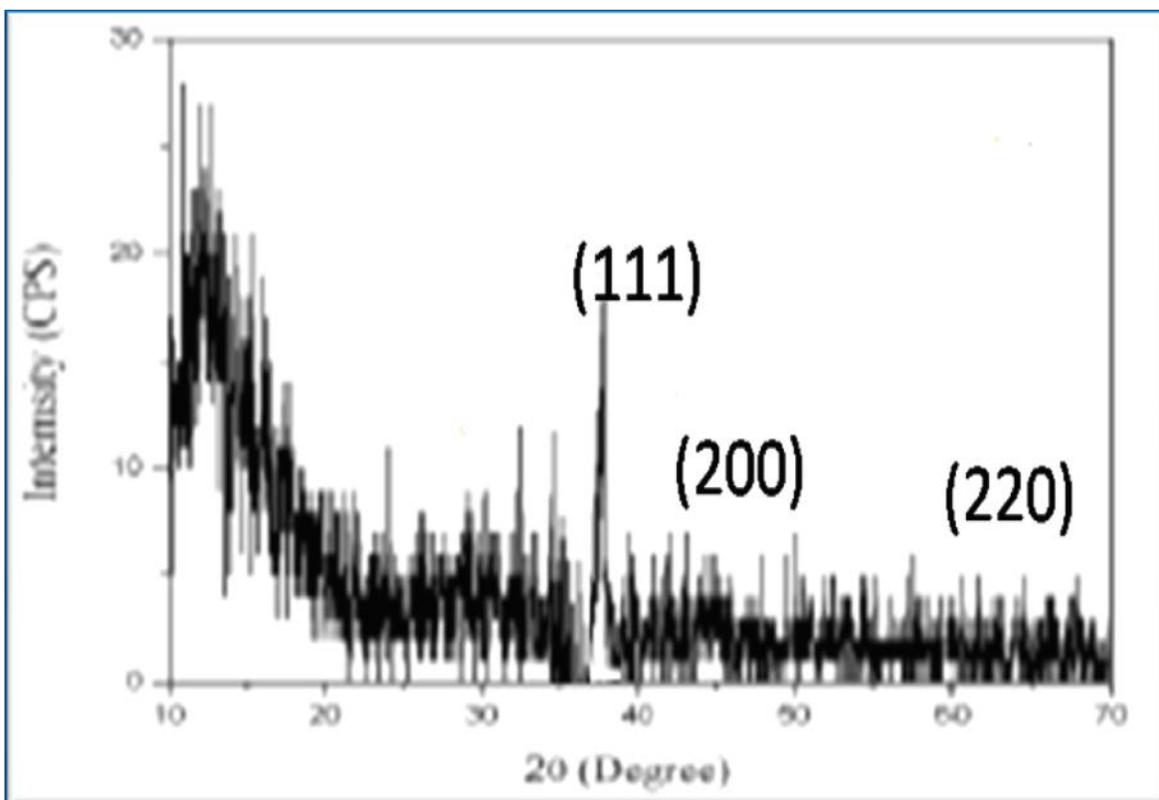


Fig. 6: XRD pattern for *A. aspera* mediated AgNPs

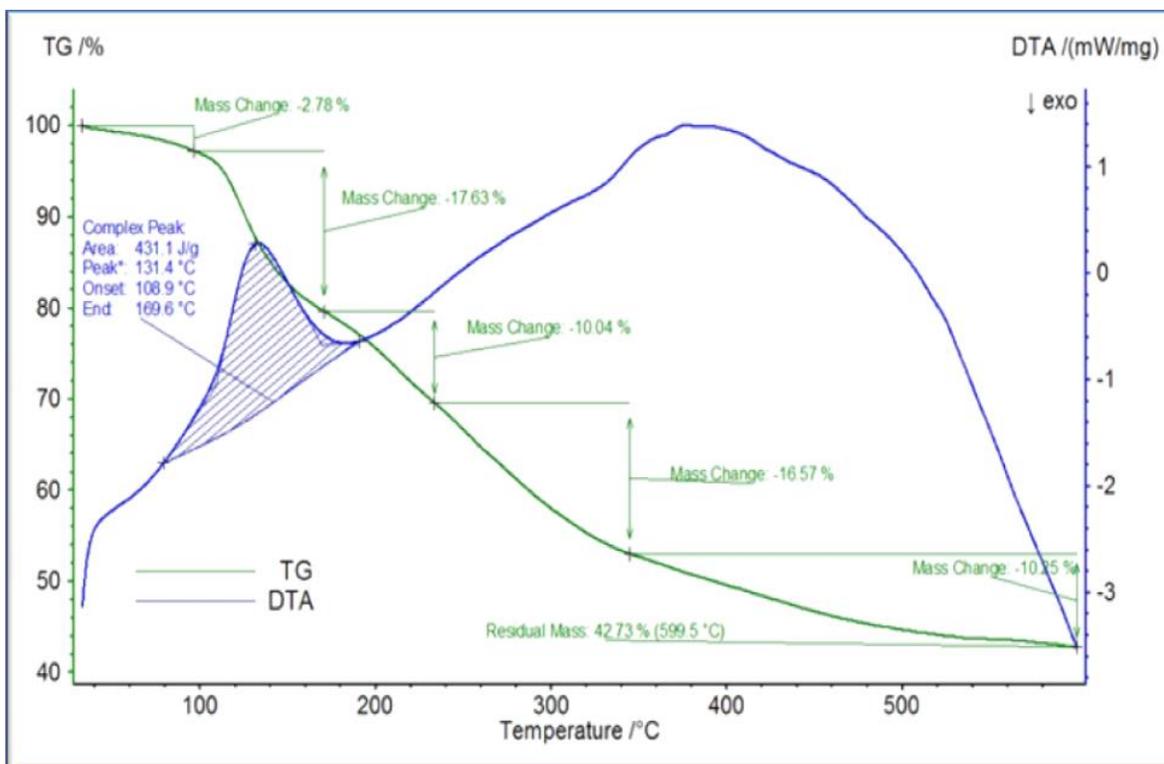


Fig. 7: TG-DTA analysis of *A. aspera* mediated AgNPs

During AgNP synthesis, the colour formation occurred within 15 min., with appearance of dark brownish black colour from pale yellowish brown colour solution. This might be due to the reduction of  $\text{Ag}^+$ , indicating the formation of AgNPs (Fig. 1). The aqueous extract-based AgNPs showed prominent peak around  $\lambda_{\text{max}}$  450 nm within 15 min. (Fig. 2) with the elevated dark brownish black colour formation. Based on colour change and UV-Vis spectral analysis, aqueous extract-based synthesized *A. aspera* AgNPs were taken for further analysis. Similarly, AgNPs synthesized by using aqueous extract of *Sargassum polycystum* showed absorbance at 430 nm [41], *Sargassum longifolium* showed absorbance peak at 460 nm [42], *Eucalyptus hybrid* [43], *Acalypha indica* [44], *Solanum tarvum* [45], *Helianthus annus* [46], and *Cassia auriculata* [47], the absorbance peaks were between 400 and 450 nm. When compared with these plants and seaweeds, AgNPs synthesized from aqueous extract of *A. aspera* were active at relatively lower wavelength.

Fig. 3 shows the FT-IR spectra of aqueous leaf extract of *A. aspera* in which the interaction of biomolecules had intensive peak at 1000 and 3500  $\text{cm}^{-1}$  (carboxylic groups) and it indicates the hydrogen bonded (O-H) stretch and the purified AgNPs at different hours shows peak shift of 3373 and 2925 of carboxylic groups indicates the O-H stretch. Earlier reports on the plant derived compounds *viz.*, polyphenols like tannic acids have emphasized their efficacy as reducing agent in the synthesis of AgNPs [48, 49]. In the present study, the band at 3524  $\text{cm}^{-1}$  is assigned for O-H stretching vibration of alcohol and phenol compounds and band observed at 1627.3  $\text{cm}^{-1}$  corresponds to N-H groups of primary amines. The band at 1034.2  $\text{cm}^{-1}$  shows that the C-O stretching vibrations of alcohols and carboxylic groups. Proteins present in the extract can bind to AgNPs through either free amino/carboxylic group in the proteins (Fig. 3A and Fig. 3B). Particularly, the peak at 1034.2  $\text{cm}^{-1}$  of the extract changed is to 1037  $\text{cm}^{-1}$  after synthesis confirming the reduction of  $\text{Ag}^+$  to AgNPs.

The morphology and size of the particles were determined by TEM and SEM. Fig.4 shows that the particles are spherical and triangular in shape well dispersed in nature and overall particle size ranges from 35.40 nm to 50.89 nm in 500 nm scale as depicted in TEM image. SEM was also used to investigate the morphology and size of the AgNPs. SEM image was recorded at different magnification and SEM image showed high density of AgNPs synthesized by *A. aspera* with the spherical and ovoid morphology (Fig. 5C). In contrary, aqueous extract

and aqueous 1mM silver nitrate solution showed aggregated morphology (Fig. 5A and Fig. 5B) and SEM image has clearly proved the bioreduction of  $\text{Ag}^+$  to  $\text{Ag}^0$  by the formation of spherical ovoid morphology. Overall the size and morphology distribution of AgNPs was found to be below 100 nm.

Elemental Ag can be seen in the graph presented by the EDX analysis in support of SEM results, which indicated the reduction of  $\text{Ag}^+$  to elemental silver (Fig. 5 D). The recent study of Khalifa *et al.* [50] stated that particle size distribution obtained by DLS was found to be mono-dispersed, or poly-dispersed ranging from 10 nm to 100 nm. XRD analysis of the NPs showed intense peak corresponding to (111), (200) and (220) Bragg's reflection based on the face centered cubic structure of AgNPs (Fig. 6). The peak corresponding to (111) plane is more intense than the other planes, suggesting that the (111) plane is in the predominant orientation. Similar results were reported by [50-51], these author's biosynthesized AgNPs using *Cinnanonum camphora* and marine algae (*Sargassum polycystum* and *Sargassum longifolium*) extract compound respectively.

TG-DTA analyses for biosynthesized AgNPs have been recorded. A ceramic crucible used for heating the sample and the analysis were carried in an atmosphere of  $\text{N}_2$  at the heating rate of 20  $^\circ\text{C}$  per minute and the temperature ranges from 30  $^\circ\text{C}$  to 600  $^\circ\text{C}$ . The TG-DTA curve of biosynthesized silver NPs is illustrated in Fig. 7. The initial mass of the material subjected to analysis was 2 mg and final mass left out after the experiment was only 42.73% (599.5 $^\circ\text{C}$ ) of the initial mass at the temperature of about 600 $^\circ\text{C}$  indicating that bulk decomposition occurred in the sample. When the biosynthesized AgNPs reached 90 $^\circ\text{C}$ , the weight loss was up to 2.78-10.25%, which is basically due to vapor or water being released. The TG-DTA analysis of AgNPs revealed that NPs was thermally stable at 90 $^\circ\text{C}$  in a nitrogen atmosphere. According to Ankamwar *et al.* [52], the thermogravimetric analysis of tamarind leaf extract reduced gold nanotriangle powder showed an initial weight loss at 125 $^\circ\text{C}$ , which is due to the presence of water molecules in the tamarind leaf extract and also there was a steady weight loss until 600  $^\circ\text{C}$ . The weight loss may due to desorption of 22 % of calcium and other bioorganic compounds present in the NPs.

### CONCLUSION:

In this paper, we have reported cost-effective and eco-friendly bio reducing method for synthesizing silver NPs using fresh leaves aqueous extract. These biologically synthesized silver NPs play a crucial role

in protecting our environment as green. UV-Vis spectroscopy revealed the surface plasmon property, while TEM and SEM with EDX image revealed the nano nature of the prepared sample. The structural analysis by XRD strongly suggests the formation of elemental silver NPs instead of their oxides in biosynthesized NPs. The XRD structural analysis of AgNPs showed that they were crystalline in nature, which might be due to the presence of bioreduction of AgNPs. The thermal stability of AgNPs was proved by TG-DTA, where AgNPs were found to be thermally stable up to 90°C and the residual mass (599.5°C), in sample was 42.73%, indicating the small amount of organic contents in synthesized sample. Therefore, the use of natural antioxidants for the synthesis of AgNPs seems to be a good alternative which could be attributed to its benign composition. The plant material responsible for the reduction and stabilization of nanoparticle needs further study including extraction and identification of the bioactive compounds presented in the extract.

#### ACKNOWLEDGEMENTS:

The authors are grateful to Central Council for Research in Ayurveda and Siddha, Chennai, for authentic identification and phytochemical analysis, Centralized Instrumentation Lab, Tamil Nadu Veterinary and Animal Science University, Chennai, for TEM analysis, Crystal Growth Center, Anna University, Chennai, for SEM with EDX, Department of Physics, Queen Mary's College, Chennai, for FTIR, NETZSCH Technologies, Chennai, for TG-DTA analysis, Department of Nuclear Physics, University of Madras, Chennai, for XRD analysis and S. Murugesan, JRF, Entomo-Pathology Lab, Forest Protective Division, IFGTB, Coimbatore, for the financial support for TEM analysis.

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